



UNITATI DE PRESIUNE

Presiuni	$\frac{N}{m^2}, Pa$	$\frac{Kgf}{m^2}, mm H_2O$	$\frac{Kgf}{cm^2}, at$	Atm	Torr, 1 mm Hg	Bar
Pascal	1	0, 10197	$1, 01972 \times 10^{-5}$	$0, 98692 \times 10^{-5}$	$7, 50064 \times 10^{-3}$	10^{-5}
Kg forta pe m^2 , mm H ₂ O	9, 80665	1	10^{-4}	$9, 67837 \times 10^{-5}$	0, 073556	$9, 80665 \times 10^{-5}$
Kg forta pe cm^2 , atmosfera tehnica (at)	$9, 80665 \times 10^4$	10^4	1	0, 967841	735, 559	0, 98006
Atmosfera fizica (atm)	$10, 1325 \times 10^4$	$1, 03323 \times 10^4$	1, 03323	1	760	$1, 33322 \times 10^{-3}$
Torr. (1 mm Hg)	133, 322	13, 5951	$13, 5951 \times 10^{-4}$	$1, 31579 \times 10^{-3}$	1	$1, 33322 \times 10^{-3}$
Bar	10^5	$1, 01972 \times 10^4$	1, 01972	0, 98692	750, 064	1